

A Computational Tool For Sensitivity Analysis in Chemical Reaction Networks Modeling Cancer Cell Signaling

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Introduction and Goal

Biological Background

Chemical Reaction Networks (CRNs)

- ▶ are a biological model capable of describing phenomena that happen inside a cell;
- ▶ model the flow of a collection of reactions;
- describe the interaction between many different cellular pathways.

Realistic CRNs consist of a huge number of chemical reactions r that involve a large number n of chemical species. In general,

CRNs translate extracellular signals into intracellular responses;

Mathematical Background

A CRN can be modeled through a **system of ODE**:

$$egin{cases} \dot{\mathrm{x}}(t) = \mathrm{Sv}(\mathrm{x}(t),\mathrm{k}^0) \ \mathrm{x}(0) = \mathrm{x}_0 \end{cases}$$

where

- $\triangleright \mathbf{x}(t) = (x_1(t), \dots, x_n(t))^T$: chemical species concentrations
- \triangleright k⁰ = $(k_1^0, \ldots, k_r^0)^T$: reaction rate constants
- \triangleright v(x(t), k⁰): reaction fluxes
- **S**: stoichiometric matrix

Our Goal: Local Sensitivity Analysis

It can be shown [5] that the equilibrium point $x^e = \lim_{t \to \infty} x(t)$ is the solution of 1 Ω

$$\begin{cases} S_2 v(x, k^0) = 0\\ Nx - c^0 = 0 \end{cases}$$

with S₂ submatrix of S. Defined $h^0 = (k^0, c^0)$
 $h^0 \rightarrow h - h^0 + Ah$

 $\neg \Pi = \Pi^{-} + \Delta \Pi$ x^e(h)

- CRNs can be altered by the presence of different types of **mutations**, causing diseases like cancer [2];
- ► Targeted **drugs** can restore CRNs' physiological behavior.

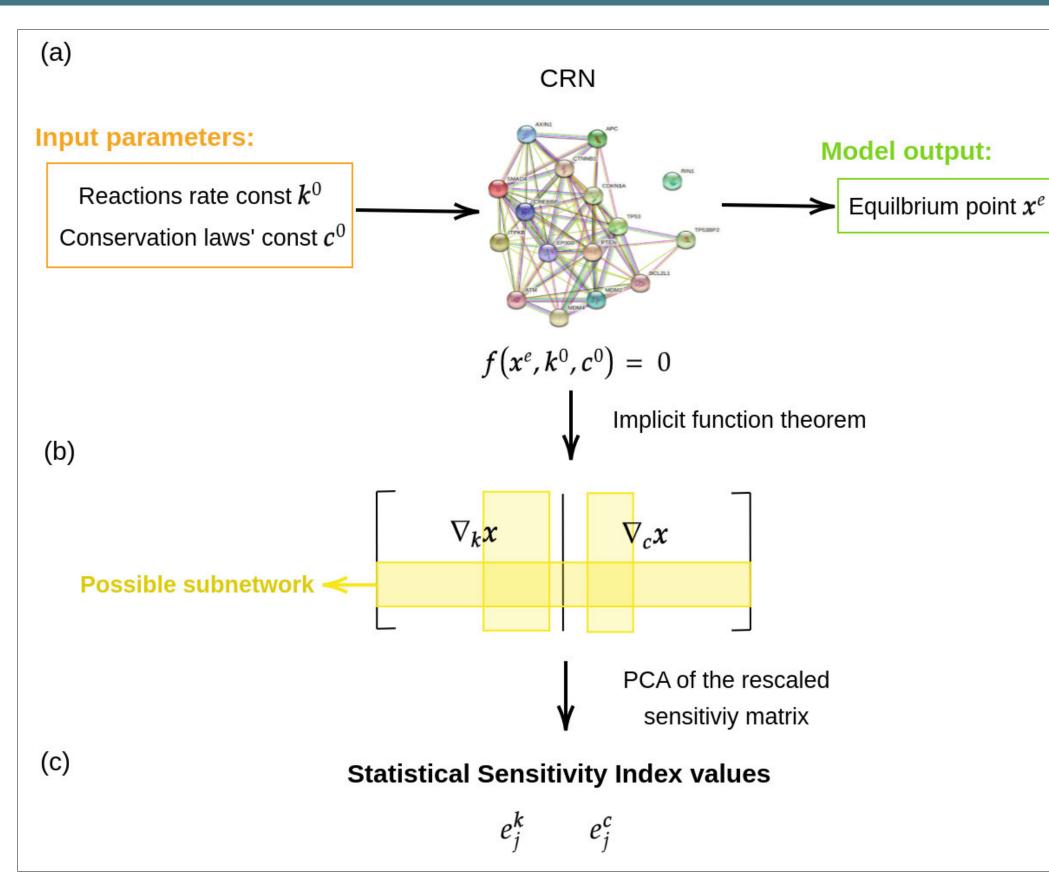
 \blacktriangleright x₀: initial condition By defining $\mathbf{N} \in \mathbb{R}^{p imes n}$ such that $\mathbf{NS} = 0$ and $\mathbf{c}^0 := \mathbf{Nx}_0$, the CRN's conservation laws are $Nx(t) = c^0$ [1].

Motivation:

▶ model reduction

drug repurposing and the search for new therapeutic targets in cancer

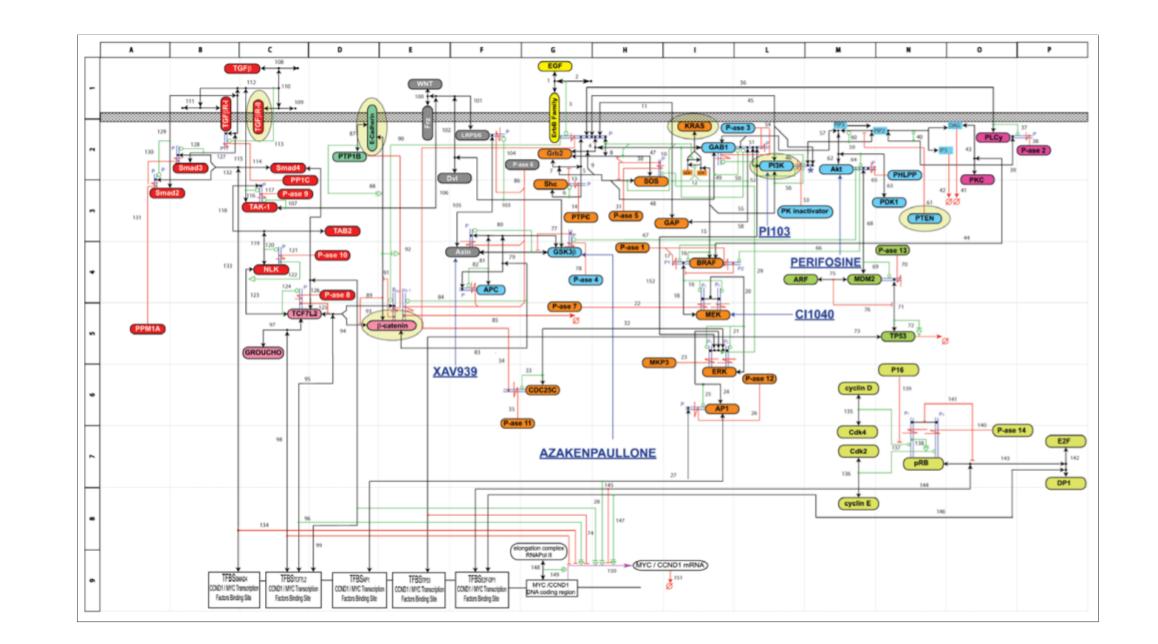
Methods



(a) Visualization of the model.

Data: Colorectal cell CRN

- Colorectal cell CRN (CRC-CRN) [3]:
- \blacktriangleright n = 419 chemical species
- ightarrow r = 850 chemical reactions
- \triangleright p = 81 conservation laws



Results: a mutated CRC-CRN

(b) By applying the Implicit Function Theorem to $f:\mathbb{R}^n imes\mathbb{R}^{r+p} o\mathbb{R}^n$

$$f(\mathbf{x},\mathbf{h}) = \begin{bmatrix} \mathbf{S}_2 \mathbf{v}(\mathbf{x},\mathbf{k}) \\ \mathbf{N}\mathbf{x} - \mathbf{c} \end{bmatrix},$$

it follows that there exists one and only one x = x(h) such that for h in a neighborhood of h^0 f(x(h), h) = 0.

The local sensitivity matrix of equilibrium with respect to parameters is obtained by computing $abla_{
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(c) The first order Taylor expansion of x(h) leads to

 $\hat{\Delta}\mathbf{x} = \boldsymbol{\mathcal{S}}\,\hat{\Delta}\mathbf{h}$,

$$\hat{\Delta} \mathbf{x} = \left[rac{x_1(\mathbf{h}) - x_1^e}{x_1^e}, \dots, rac{x_n(\mathbf{h}) - x_n^e}{x_n^e}
ight]^T, \quad \hat{\Delta} \mathbf{h} = \left[rac{h_1 - h_1^0}{h_1^0}, \dots, rac{h_{r+p} - h_{r+p}^0}{h_{r+p}^0}
ight]^T, \ oldsymbol{\mathcal{S}}_{ij} = rac{h_j^0}{x_i^e} (
abla_{\mathbf{x}}(\mathbf{h}^0))_{ij}$$

From the definition of the semi-positive quadratic form [4]

$$Q = \hat{\Delta} \mathbf{x}^T \hat{\Delta} \mathbf{x} = \hat{\Delta} \mathbf{h}^T \boldsymbol{\mathcal{S}}^T \boldsymbol{\mathcal{S}} \hat{\Delta} \mathbf{h}$$

a Principal Component Analysis of $\mathcal{S}^T \mathcal{S}$ can be computed to find the Statistical Sensitivity Indexes (SSI)

$$e_j^{\mathrm{h}} = rac{\sum\limits_{a=1}^{r+p} \lambda_a \, u_{ja}^2}{\sum\limits_{a=1}^{r+p} \lambda_a},$$

Considered mutations: (a) GoF of KRAS; (b) LoF of APC; (c) LoF of SMAD4; (d) LoF of TP53.

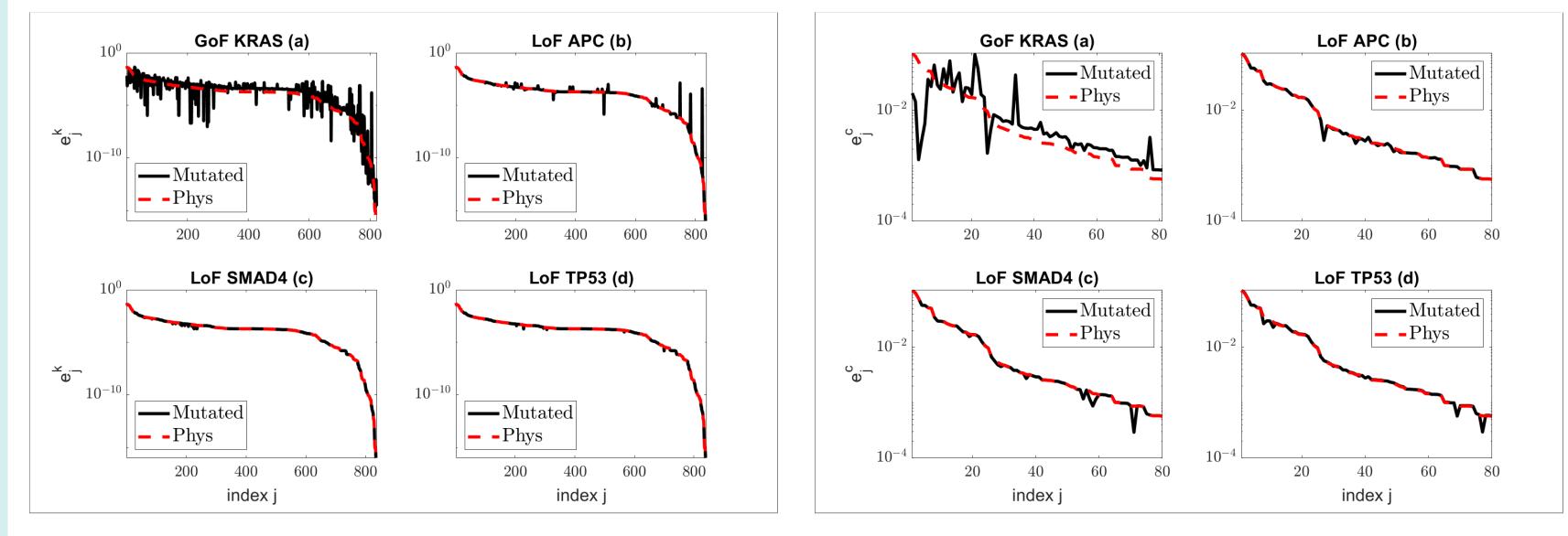


Figure SSI values for all rate constants (left panel) and for all conservation laws' constants (right panel) in the case of four mutated networks (black solid line) with respect to the SSI values of the physiological CRN (red dashed line).

Reference #	Reaction	$\delta e^{ m k}_{j}$
R41	$Ras + GDP \rightarrow Ras_GDP$	1.91e+07
R294	$ERBP_ShP_G_S \rightarrow ERBP_ShP_G + SOS$	2.91e+04
R49	$RP_ShP_G_S_Ras + GTP \longrightarrow RP_ShP_G_S_Ras_GTP$	2.88e+04
R412	$ERB3P_Sh \to ERB3P + Shc$	2.86e+04
R415	$ShP + ERB3P \rightarrow ERB3P_ShP$	1.23e+04
R297	$ERBP_ShP + G_S \longrightarrow ERBP_ShP_G_S$	8.64e+03
R302	$ERBP_ShP_G_S_Ras + GDP \rightarrow ERBP_ShP_G_S_Ras_GDP$	5.78e+02
R57	$RP_G_S_Ras_GDP \longrightarrow RP_G_S_Ras + GDP$	5.78e+02
R420	$ERB3P_ShP_G + SOS \rightarrow ERB3P_ShP_G_S$	5.78e+02
R52	$RP_ShP_G_S + Ras_GTP \rightarrow RP_ShP_G_S_Ras_GTP$	4.95e+02

where $\lambda_1,..., \lambda_{r+p}$, with $\lambda_1 \geq \lambda_2 \geq ... \geq \lambda_{r+p} \geq 0$, and $\mathbf{u}_1,..., \mathbf{u}_{r+p}$ are the eigenvalues and the corresponding normalized eigenvectors of $\mathcal{S}^{\hat{T}}\mathcal{S}$, respectively.

Mutated CRN

- ▶ A Loss of Function mutation (LoF) nullifies the concentration of a chemical species [1].
- \triangleright remove from S, x and N the entries referred to the chemical species that don't appear in the network anymore;
- ▶ remove from S and v the entries referred to the reactions that don't happen in the network anymore.
- ▶ A Gain of Function mutation (GoF) increases the concentration of a chemical species [1].
- \blacktriangleright remove from S and v the entries referred to the reactions that don't happen in the network anymore.

Table Reactions whose sensitivity indexes are mostly effected by the GoF mutation of KRAS. The third column contains the relative differences of e_i^k between the mutated and the physiological CRC-CRN.

References

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